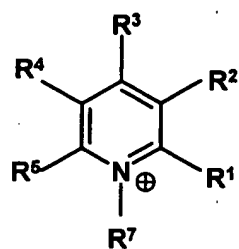


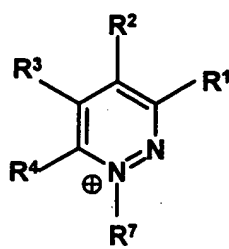
## AMENDMENTS TO THE CLAIMS

We claim:

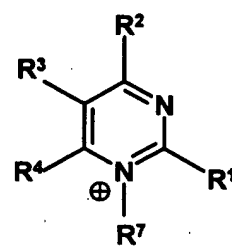
- 5
1. (currently amended) A process for preparing haloalkanes ~~by reaction of~~ comprising  
10 reacting alcohol with hydrogen halide, wherein the reaction of the alcohol with the  
hydrogen halide occurs in the presence of an ionic liquid at a temperature which is above  
100°C for at least part of the time and, at least at the time of commencement of the  
reaction, the water content is not more than 25 mol% based on the amount of ionic liquid,  
where the ionic liquid is not octyltrimethylammonium chloride.
  2. (original) A process as claimed in claim 1, wherein part or all of the hydrogen halide is  
15 passed into the mixture comprising the alcohol and the ionic liquid at temperatures below  
100°C and the mixture formed by the addition is heated to temperatures above 100°C for  
part of the time.
  3. (currently amended) A process as claimed in claim 1 ~~or 2~~, wherein the reaction is carried  
out at from 110°C to 150°C.
  - 20 4. (currently amended)) A process as claimed in ~~any of claims~~ claim 1 ~~to 3~~ wherein HCl or  
HBr is used as hydrogen halide.
  - 25 5. (currently amended) A process as claimed in ~~any of claims~~ claim 1 ~~to 4~~, wherein an  
alcohol selected from the group consisting of sec-butanol, isobutanol, 2-ethylhexanol, 2-  
propylheptanol, isononanol, cyclohexanol, cyclopentanol, glycol, 1,3-propanediol, 1,4-  
butanediol, 1,5-pentanediol, 1,6-hexanediol, neopentyl glycol, trimethylolpropane,  
pentaerythritol, glycerol, trimethylolethane, 1,2-propanediol, 1,2-butanediol, 2,3-  
butanediol, allyl alcohol, propargyl alcohol, diethylene glycol and triethylene glycol, in  
30 particular selected from the group consisting of 1,6-hexanediol, 1,5-pentanediol, 1,4-  
butanediol, 1,3-propanediol, glycol, allyl alcohol and propargyl alcohol, is used.
  6. (currently amended) A process as claimed in claim 1, wherein an ionic liquid which  
comprises one each of the following cations and anions,  
35  
where the cation is selected from the group consisting of:



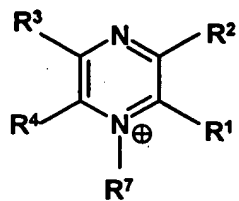
(a)



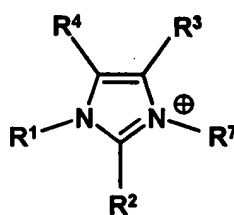
(b)



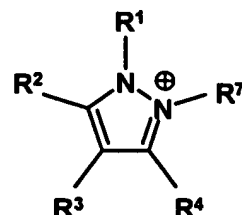
(c)



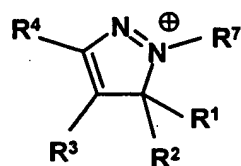
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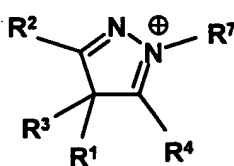
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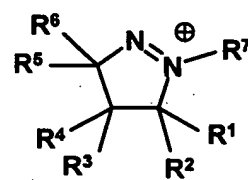
(f)



(g)

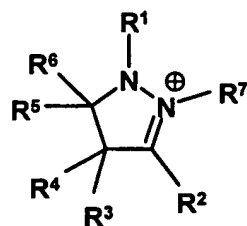


(h)

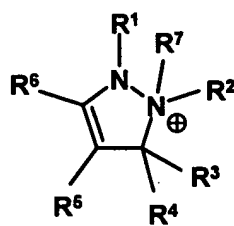


(i)

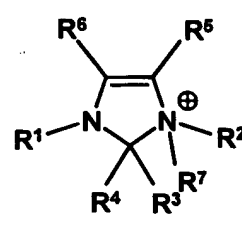
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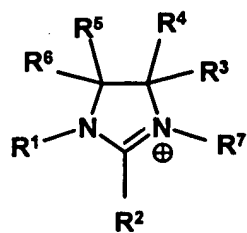
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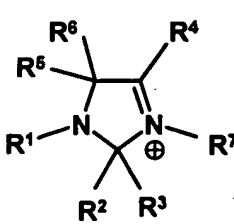
(k)



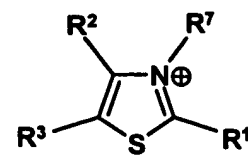
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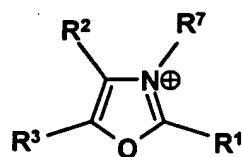
(m)



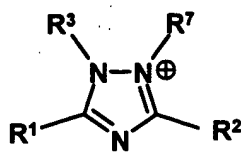
(n)



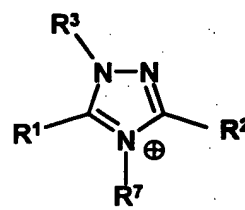
(o)



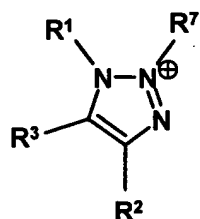
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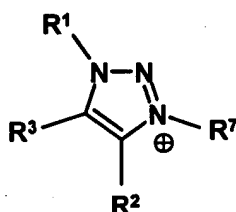
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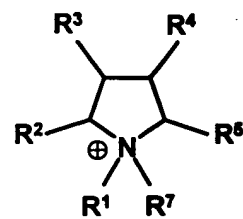
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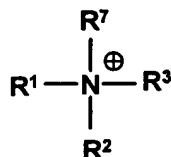
(s)



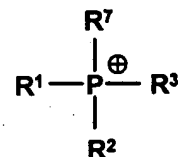
(t)



(u)



(v)



(w)

5

and also oligomers and polymers in which these structures are present,

10

where

15

$R^1$ ,  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^5$ ,  $R^6$  and  $R^7$  are each, independently of one another, hydrogen,  $C_1$ - $C_{18}$ -alkyl,  $C_2$ - $C_{18}$ -alkyl which may be interrupted by one or more oxygen and/or sulfur atoms and/or one or more substituted or unsubstituted imino groups,  $C_6$ - $C_{12}$ -aryl,  $C_5$ - $C_{12}$ -cycloalkyl or a five- or six-membered, oxygen-, nitrogen- and/or sulfur-containing heterocycle or two of them together form an unsaturated, saturated or aromatic ring which may be interrupted by one or more oxygen and/or sulfur atoms and/or one or more substituted or unsubstituted imino groups, where the radicals mentioned may each be, independently of one another, monosubstituted or polysubstituted by functional groups, aryl, alkyl, aryloxy, alkyloxy, halogen and/or heterocycles, with in the case of the ammonium ions (v),  $R^1$ ,  $R^2$ ,  $R^3$  and  $R^7$  not all being an unsubstituted alkyl radical,

20

$R^7$  can also be  $C_1$ - $C_{18}$ -alkyloyl,  $C_1$ - $C_{18}$ -alkyloxycarbonyl,  $C_5$ - $C_{12}$ -cycloalkylcarbonyl or  $C_6$ - $C_{12}$ -aryloyl, where the radicals mentioned may each be, independently of one another,

monosubstituted or polysubstituted by functional groups, aryl, alkyl, aryloxy, alkyloxy, halogen and/or heterocycles and

functional groups are: carboxyl, carboxamide, hydroxyl, amino, C<sub>1</sub>-C<sub>4</sub>-alkylamino, di(C<sub>1</sub>-C<sub>4</sub>-alkyl)amino, C<sub>1</sub>-C<sub>4</sub>-alkyloxycarbonyl, cyano or C<sub>1</sub>-C<sub>4</sub>-alkyloxy,

and the anion is selected from the group consisting of:

halides, alkylcarboxylate, tosylate, sulfonate, dialkylphosphate, bis(trifluoromethylsulfonyl)imide, trifluoroacetate, triflate, sulfate, hydrogensulfate, methylsulfate, ethylsulfate, sulfite, hydrogensulfite, chloroaluminates, bromoaluminates, nitrite, nitrate, chlorocuprate, phosphate, hydrogenphosphate, dihydrogenphosphate, carbonate and hydrogencarbonate,

is used.

7. (currently amended) A process as claimed in any of claims claim 1 to 6, wherein an ionic liquid which comprises one each of the following cations and anions,

where the cation is selected from the group consisting of:

1-methylimidazolium, 1-ethylimidazolium, 1-propylimidazolium, 1-butylimidazolium, 2-ethylpyridinium, 1-ethyl-3-methylimidazolium, 1-n-butyl-3-ethylimidazolium, 1,2-dimethylpyridinium, 1-methyl-2-ethylpyridinium, 1-methyl-2-ethyl-6-methylpyridinium, N-methylpyridinium, 1-butyl-2-methylpyridinium, 1-butyl-2-ethylpyridinium, 1-butyl-2-ethyl-6-methylpyridinium, N-butylpyridinium, 1-butyl-4-methylpyridinium, 1,3-dimethylimidazolium, 1,2,3-trimethylimidazolium, 1-n-butyl-3-methylimidazolium, 1,3,4,5-tetramethylimidazolium, 1,3,4-trimethylimidazolium, 1,2-dimethylimidazolium, 1-butyl-2,3-dimethylimidazolium, 3,4-dimethylimidazolium, 2-ethyl-3,4-dimethylimidazolium, 3-methyl-2-ethylimidazolium, 3-butyl-1-methylimidazolium, 3-butyl-1-ethylimidazolium, 3-butyl-1,2-dimethylimidazolium, 1,3-di-n-butylimidazolium, 3-butyl-1,4,5-trimethylimidazolium, 3-butyl-1,4-dimethylimidazolium, 3-butyl-2-methylimidazolium, 1,3-dibutyl-2-methylimidazolium, 3-butyl-4-methylimidazolium, 3-butyl-2-ethyl-4-methylimidazolium, 3-butyl-2-ethylimidazolium, 1-methyl-3-octylimidazolium and 1-decyl-3-methylimidazolium,

and the anion is selected from the group consisting of:

halides, acetate, methanesulfonate, tosylate, sulfate, hydrogensulfate, phosphate, hydrogenphosphate, dihydrogenphosphate, dialkylphosphate and bis(trifluoromethylsulfonyl)imide,

is used.

8. (currently amended) A process as claimed in ~~any of claims~~ claim 1 to 7, wherein an ionic  
5 liquid selected from the group consisting of: 1-methylimidazolium chloride, 1-methylimidazolium bromide, 1-methylimidazolium fluoride, 1-methylimidazolium iodide, 1-methylimidazolium hydrogensulfate, 1-methylimidazolium sulfate, 1-methylimidazolium methanesulfonate, 1-methylimidazolium tosylate, 1-methylimidazolium diethylphosphate, 1-ethylimidazolium chloride, 1-ethylimidazolium  
10 bromide, 1-ethylimidazolium fluoride, 1-ethylimidazolium iodide, 1-ethylimidazolium hydrogensulfate, 1-ethylimidazolium sulfate, 1-ethylimidazolium methanesulfate, 1-ethylimidazolium tosylate, 1-ethylimidazolium diethylphosphate, 1-propylimidazolium chloride, 1-propylimidazolium bromide, 1-propylimidazolium fluoride, 1-propylimidazolium iodide, 1-propylimidazolium hydrogensulfate, 1-propylimidazolium  
15 sulfate, 1-propylimidazolium methanesulfate, 1-propylimidazolium tosylate, 1-propylimidazolium diethylphosphate, 1-butylimidazolium chloride, 1-butylimidazolium bromide, 1-butylimidazolium fluoride, 1-butylimidazolium iodide, 1-butylimidazolium hydrogensulfate, 1-butylimidazolium sulfate, 1-butylimidazolium methanesulfonate, 1-butylimidazolium tosylate, 1-butylimidazolium diethylphosphate, 2-ethylpyridinium  
20 chloride, 2-ethylpyridinium bromide, 2-ethylpyridinium iodide, 2-ethylpyridinium hydrogensulfate, 2-ethylpyridinium sulfate, 2-ethylpyridinium methanesulfonate, 2-ethylpyridinium tosylate, 2-ethylpyridinium diethylphosphate, 1-ethyl-3-methylimidazolium chloride, 1-ethyl-3-methylimidazolium bromide, 1-ethyl-3-methylimidazolium fluoride, 1-ethyl-3-methylimidazolium iodide, 1-ethyl-3-methylimidazolium  
25 hydrogensulfate, 1-ethyl-3-methylimidazolium sulfate, 1-ethyl-3-methylimidazolium methanesulfonate, 1-ethyl-3-methylimidazolium tosylate, 1-ethyl-3-methylimidazolium diethylphosphate, 1-n-butyl-3-methylimidazolium chloride, 1-n-butyl-3-methylimidazolium bromide, 1-n-butyl-3-methylimidazolium fluoride, 1-n-butyl-3-methylimidazolium iodide, 1-n-butyl-3-methylimidazolium  
30 hydrogensulfonate, 1-n-butyl-3-methylimidazolium sulfate, 1-n-butyl-3-methylimidazolium methanesulfonate, 1-n-butyl-3-methylimidazolium tosylate, 1-n-butyl-3-methylimidazolium diethylphosphate, 1-n-butyl-3-ethylimidazolium chloride, 1-n-butyl-3-ethylimidazolium bromide, 1-n-butyl-3-ethylimidazolium fluoride, 1-n-butyl-3-ethylimidazolium iodide, 1-n-butyl-3-ethylimidazolium hydrogensulfate, 1-n-butyl-3-ethylimidazolium sulfate, 1-n-butyl-3-ethylimidazolium methanesulfonate, 1-n-butyl-3-ethylimidazolium tosylate and 1-n-butyl-3-ethylimidazolium diethylphosphate,

~~in particular selected from the group consisting of: 1-methylimidazolium chloride,  
1-methylimidazolium bromide, 1-methylimidazolium hydrogensulfate, 2-ethylpyridinium  
40 chloride, 2-ethylpyridinium bromide, 2-ethylpyridinium hydrogensulfate, 1-ethyl-3~~

~~methylimidazolium chloride, 1-ethyl-3-methylimidazolium bromide and 1-ethyl-3-methylimidazolium hydrogensulfate,~~

is used.

5

9. (currently amended) A process as claimed in ~~any of claims 1 to~~ claim 8, wherein the haloalkane ionic liquid is isolated selected from the ~~reaction mixture by distillation group consisting of: 1-methylimidazolium chloride, 1-methylimidazolium bromide, 1-methylimidazolium hydrogensulfate, 2-ethylpyridinium chloride, 2-ethylpyridinium bromide, 2-ethylpyridinium hydrogensulfate, 1-ethyl-3-methylimidazolium chloride, 1-ethyl-3-methylimidazolium bromide and 1-ethyl-3-methylimidazolium hydrogensulfate.~~
10. (currently amended) A ~~process as claimed in any of claims 1 to 9, wherein the ionic liquid has a melting point of less than 150°C, in particular less than 100°C. A process as~~ claimed in claim 1, wherein the haloalkane is isolated from the reaction mixture by distillation.
11. (currently amended) A process as claimed in ~~any of claims claim 1 to 10, wherein the reaction is carried out in the presence of from 1 to 3 mol of ionic liquid per mol of OH group to be reacted in the alcohol.~~ has a melting point of less than 150°C.
12. (currently amended) A process as claimed in ~~any of claims 1 to claim 11, wherein the reaction melting point is carried out in the absence of water or the substantial absence of water at the time of commencement of the reaction~~ less than 100°C.
13. (currently amended) A ~~process as claimed in any of claims 1 to 12, wherein the water liberated in the reaction is continuously removed, in particular distilled off. A process as~~ claimed in claim 1, wherein the reaction is carried out in the presence of from 1 to 3 mol of ionic liquid per mol of OH group to be reacted in the alcohol.
14. (currently amended) A process as claimed in claim 13, wherein the ~~water content in the reaction is carried out in the absence of water or the alcohol with substantial absence of water at the hydrogen halide is not more than 25 mol%, preferably not more than 20 mol%, more preferably not more than 10 mol%, particularly preferably not more than 5 mol%, based on time of commencement of the amount of ionic liquid, over the entire reaction time.~~
15. (currently amended) A process as claimed in claims 1 to 14, wherein in the case of the reaction of alcohols having more than one OH group per molecule, all OH groups are replaced by halogen.

16. (new) A process as claimed in claim 1, wherein the water liberated in the reaction is continuously removed.
- 5 17. (new) A process as claimed in claim 16, wherein the water liberated in the reaction is distilled off.
- 10 18. (new) A process as claimed in claim 16, wherein the water content in the reaction of the alcohol with the hydrogen halide is not more than 25 mol%, based on the amount of ionic liquid, over the entire reaction time.
19. (new) A process as claimed in claim 18, wherein the water content is not more than 5 mol%, based on the amount of ionic liquid.